isomerization makes such a process very attractive. It is noteworthy that although the proton transfer in the conversion of I to II (eq 1) involves a 1,8 and a 1,13 shift, respectively, no competition of the simple vinyl epoxide isomerization to enones occurs.¹¹ This type of cyclization holds promise of being a general solution to normally unfavorable ring sizes. Furthermore, the juxtaposition of functionality in the specific cases chosen offers an opportunity for further elaboration toward natural products such as the polyene macrolides.¹⁵ A simple illustration from 13 to muscone can be envisioned by conjugate addition of lithium dimethylcuprate followed by reductive desulfonylation.¹⁶

Acknowledgment. We thank the National Science Foundation and the General Medical Sciences Institute of the National Institutes of Health for their generous support of our programs. We are grateful to Rohm and Haas for a generous gift of the polystyrene resin and Englehardt Industries and Johnson Matthey for a generous supply of palladium chloride.

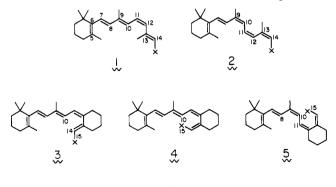
Registry No. 1 ($R = CO_2CH_3$) Na, 82902-78-1; 1 ($R = PhSO_2$) Na, 82917-43-9; 2, 82902-79-2; 3, 82902-80-5; 4, 82902-81-6; 5, 82917-44-0; 6, 82902-82-7; 7, 82902-83-8; 8, 82902-84-9; 9, 82902-85-0; 10, 82902-86-1; 11, 82902-87-2; 12, 82902-88-3; 13, 82902-89-4; 14, 82902-90-7; ethyl 8-bromo-2-octenoate, 82902-91-8; ethyl 13-bromo-2-tridecenoate, 82902-92-9; 8-bromo-2,3-epoxyoctanol, 82902-93-0; 13-bromo-2,3-epoxytridecanol, 82902-94-1; 2-(5-bromopentyl)-3-ethenyloxirane, 82902-95-2; 2-(10-bromodecyl)-3-ethenyloxirane, 82902-96-3; 6-bromohexanol, 4286-55-9; 11-bromoundecanol, 1611-56-9; methyl 1-phenylsulfonyl-3cyclononene-1-carboxylate, 70255-45-7; 4,4-diphenylsulfonylcyclononene, 70255-47-9; methyl 1-phenylsulfonyl-2-ethenylcycloheptanecarboxylate, 70255-46-8; 1,1-diphenylsulfonyl-2-ethenylcycloheptane, 70255-48-0; (Ph₃P)₄Pd, 14221-01-3; dppe, 1663-45-2.

12-s-Cis-Locked Retinoids (Vitamin A): Synthesis and **Novel Spectral Properties**

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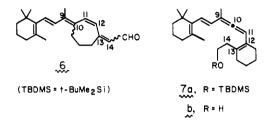
The chromophoric group of the visual pigment rhodopsin, 11cis-retinal (1a),¹ is believed to exist in solution as an equilibrium



b, ×=CH2OH × = CHO ;

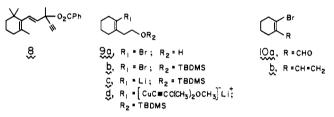
mixture of twisted 12-s-cis and 12-s-trans conformers.² The observation that 1a and the corresponding alcohol 1b exhibit UV

maxima at 377 and 319 nm,³ respectively, is indicative of a highly distorted chromophore for each.³ We recently made the unusual observation that the more hindered 9-cis,11-cis,13-cis-retinal (2a) exhibits its main maximum (302 nm) to the blue of the corresponding alcohol 2b (306 nm).^{4a} The ¹H NMR data indicated that a high degree of distortion exists about the C(12)-C(13) single bond.^{4a} In order to further evaluate this effect, this communication describes the synthesis and spectral properties of highly twisted 12-s-cis-locked retinal analogues possessing 11-cis (3a), 11cis,13-cis (4a), and 9-cis,11-cis,13-cis (5a) geometries. This study nicely complements that of Nakanishi and co-workers on twisted 12-s-trans-locked retinals 6, which exhibit relatively normal



electronic spectra ($\lambda_{max} > 350$ nm) for both the 11-cis and 9-cis,11-cis,13-cis geometric isomers.⁵ The synthetic route used for preparing 3-5 entails the [1,5] sigmatropic hydrogen shift of the vinylallene 7b as a key step, a process specific for producing 11-cis-retinoids.⁴ This process should have also produced a 9cis,11-cis isomer, but the latter apparently undergoes further previously unrecognized^{4a} pericyclic transformations. This study therefore also provides further insight into the stereochemical course of vinvlallene rearrangements.

The vinylallene 7a was obtained by the coupling of the propargyl benzoate 8 and the vinyl cuprate 9d,^{4,6} which was prepared as



follows. Reaction of 10a (prepared in 52% yield from cyclohexanone)⁷ under Wittig conditions (ether, Ph₃PCH₃Br, *n*-BuLi; 54%) gave the bromodiene 10b. Hydroboration-oxidation (THF, 9-BBN; H₂O₂, NaOH, H₂O-CH₃OH; 85%)⁸ followed by protection (TBDMSCl, imidazole, DMF; 84%)⁹ afforded the bromosilyl ether 9b. Lithiation (2.1 equiv of t-BuLi, ether, -78 °C, 4 h) of **9b** followed by reaction with $CuC \equiv C - C(CH_3)_2 OCH_3^{10}$ afforded the mixed cuprate 9d, which was reacted with 8 to afford silvl ether 7a. Deprotection $((n-Bu)_4NF, THF, room temperature,$ 4.5 h)⁹ of **7a** followed by high-pressure liquid chromatography (HPLC) (15% ethyl acetate-skellysolve B) afforded pure vinyl-

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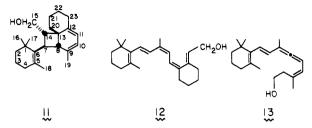
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allene 7b (43% based on 9b).11

The thermally induced rearrangement of the vinylallenol 7b $(10^{-3} \text{ M in purified hexanes at reflux, } \sim 69 \text{ °C, 4 h})$ followed by preparative HPLC afforded four components (82% mass balance) in the following order of elution: a substance characterized as 11 (12%) and the 12-s-cis-locked retinols, 11-cis, 13-cis-4b (33.1%),



11-cis-3b (13.5%), and 9-cis,11-cis,13-cis-5b (22.8%). Each of the four thermolysis products was individually subjected to the conditions of thermolysis (refluxing hexanes, 4 h) and found to be unchanged by ¹H NMR and HPLC analyses.

The observation of 11 as one of the thermolysis products of 7b is of particular interest since no retinol possessing the 9-cis,11-cis geometry was observed as a product in the thermolysis of the previously reported vinylallenol 13.4a Structure 11 is proposed on the basis of spectral analysis¹¹ and on the reasonable mechanistic hypothesis that it can be derived from further rearrangement of the putative 9-cis, 11-cis-isomer 12. Examination of 12 reveals in it the presence of trans, cis, cis, trans-octatetraene mojety. Such tetraenes are known to undergo extraordinarily facile eight-electron conrotatory electrocyclization to cycloocta-1,3,5-trienes, which further electrocyclize to bicyclo[4.2.0]octa-2.4-dienes in a six-electron disrotatory manner.¹² A similar series of tandem electrocyclizations should result in the rearrangement of the putative intermediate 12 to 11.

The retinol analogues 3b, 4b, and 5b are considered to be of the 11-cis geometrical series on the basis of their method of synthesis via the thermal [1,5] sigmatropic rearrangement. The 13-cis geometry for 4b and 5b was based on observation of an NOE between H₁₅ and H₁₀ in these isomers while no NOE was observed between these protons in 3b. Compound 5b was assigned the 9-cis geometry on the basis of deshielding of H_8 (~0.6 ppm) as a result of interaction with H₁₁.^{11,13} The retinols were individually oxidized (MnO₂, low-boiling petroleum ether, 0 °C, 1 h; $\sim 80\%$) to the corresponding retinals, and the spectral data for these compounds were also in accord with the assigned geometries.11

The electronic absorption spectra of retinals generally exhibit prominent α bands (~360 nm) and weak β bands (~280 nm).¹⁴ In striking contrast, the absorption spectra of the 12-s-cis-locked retinal analogues show marked enhancements of the β bands relative to the α bands,¹¹ a phenomenon also exhibited by 9-cis,11-cis,13-cis-retinal.^{4a} Interestingly, the corresponding 12s-trans-locked analogues exhibit maxima in the α region.⁵ It thus appears that the appearance of a strong absorption at ~ 300 nm is a consequence of distortion of the chromophore into a twisted 12-s-cis conformation. The data also suggest that 9-cis,11cis, 13-cis-retinal (2a) exists predominantly in a twisted 12-s-cis conformation.

This paper further defines the scope of the vinylallene method for synthesizing 11-cis-retinoids. Apparently, the application of this approach to 9,10-allenes results in stereospecific production of 11-cis isomers, but there is little control of the stereochemistry of the lateral double bonds (Δ^9 and Δ^{13}) and the 9-cis,11-cis isomers are not stable to the thermal conditions. Future research will be directed toward addressing these problems. Despite these shortcomings, the method still allows rapid access to certain of the hindered 11-cis-retinoid analogues in sufficient quantities for further study. Recent reports^{15,16} ascribing high activity in the chemoprophylaxis of epithelial cancer to retinoid analogues possessing 12-s-cis topologies make 3-5 attractive candidates for related biological evaluation.

Acknowledgment. The National Institutes of Health (USPHS Grant EY-02452 and NCI Contract CP-05715) provided financial support for this project. We also acknowledge the National Science Foundation Midwest Center for Mass Spectrometry for mass spectra, the Southern California Regional NMR Facility (supported by NSF Grant No. CHE79-16324) for 500-MHz spectra, and BASF (Ludwigshafen) for chemicals.

Registry No. 3a, 83043-75-8; 3b, 83043-76-9; 4a, 83043-77-0; 4b, 83043-78-1; **5a**, 83043-79-2; **5b**, 83043-80-5; **7a**, 83043-81-6; **7b**, 83043-82-7; **8**, 74723-00-5; **9b**, 83043-83-8; **9d**, 83043-86-1; **10a**, 38127-47-8; 10b, 83043-84-9; 11, 83060-55-3; 12, 83043-85-0.

Supplementary Material Available: Spectral data for 7b, 3ab, 4ab, 5ab, and 11 (2 pages). Ordering information is given on any current masthead page.

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Electrocyclization of 1-Allenyldienes: Novel Synthesis of Drimatrienes and Related trans-Decalins

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Vinylallenes of the type 1 are useful intermediates for preparing polyenes such as vitamin D^1 and 11-cis-retinoids,² wherein a key step is a $C_6 \rightarrow C_2$ hydrogen migration, a thermal suprafacial [1,5]sigmatropic shift.^{3,4} In order to further define the scope of this allene approach in organic synthesis, we considered the diene-allene⁵ of the type 2; it was initially assumed that it would undergo an exceptionally rapid $C_8 \rightarrow C_2$ antarafacial [1,7]sig-

⁽¹¹⁾ All new compounds were characterized by ¹H and ¹³C NMR, IR, and low- and high-resolution mass spectra and by UV as appropriate (see sup-plementary material section). The UV data (95% EtOH) for the retinals plementary material section). The UV data (95% EICH) for the retinals 3a-5a and retinols 3b-5b are as follows: $(3a) \lambda_{max} 358 \text{ nm}$ (sh, $\epsilon 9500$), 288 nm ($\epsilon 19 300$), 251 nm ($\epsilon 19 500$), 235 nm (sh, $\epsilon 17 600$); (4a) $\lambda_{max} 357 \text{ nm}$ (sh, $\epsilon 4900$), 300 nm ($\epsilon 27 200$), 236 nm ($\epsilon 20 300$); (5a) $\lambda_{max} 362 \text{ nm}$ (sh, $\epsilon 3200$), 299 nm ($\epsilon 23 000$), 233 nm ($\epsilon 18 800$); (3b) $\lambda_{max} 308.5 \text{ nm}$ ($\epsilon 19 700$), 230 nm ($\epsilon 24 400$), 223 nm ($\epsilon 9400$), 230 nm ($\epsilon 24 400$), 233 nm ($\epsilon 28 800$), 230 nm ($\epsilon 11 000$); (5b) $\lambda_{max} 303 \text{ nm}$ ($\epsilon 24 400$), 223 nm ($\epsilon 9400$).

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